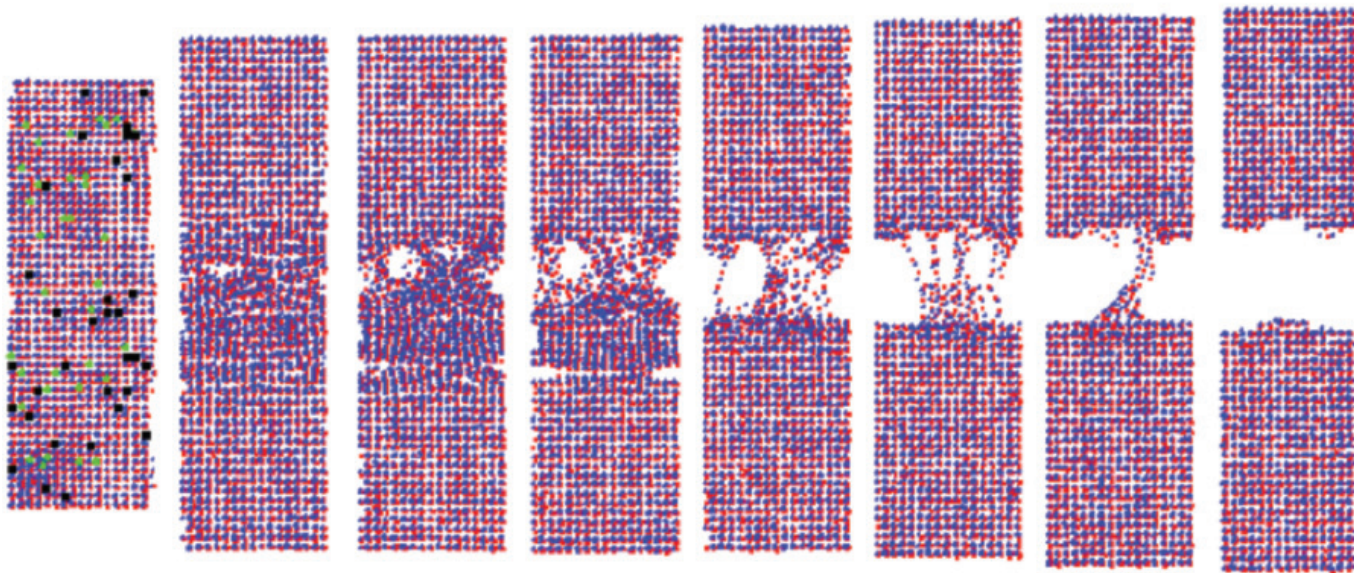


Modeling of Laser Ablation of LiF - Influence of Defects

H. M Urbassek, Y. Cherednikov
Physics Dept., University of Kaiserslautern, Germany

N. A. Inogamov
Landau Institute, Russian Academy of Science



Theoretical tools

- **Molecular dynamics**

Solve Newton 's equations.

Advantages:

- Input: only interatomic forces, nowadays available for many materials
- as realistic as possible
 - for many-body simulations
 - for thermal nonequilibrium situations
- easy visualization / animation: appeals to imagination

Disadvantages:

- computationally slow
- cannot handle time scales & 1 ns
- cannot handle space scales & 100 nm
[1 μm]



Isaac Newton (1643 – 1727)
1687: Philosophiae Naturalis
Principia Mathematica

Outline

- **Two-temperature model / MD for metals**
- Two-temperature model / MD for LiF
- Melting and spallation of thin LiF films
- defects
- swift-ion tracks in LiF

Two-temperature model + MD for metals:

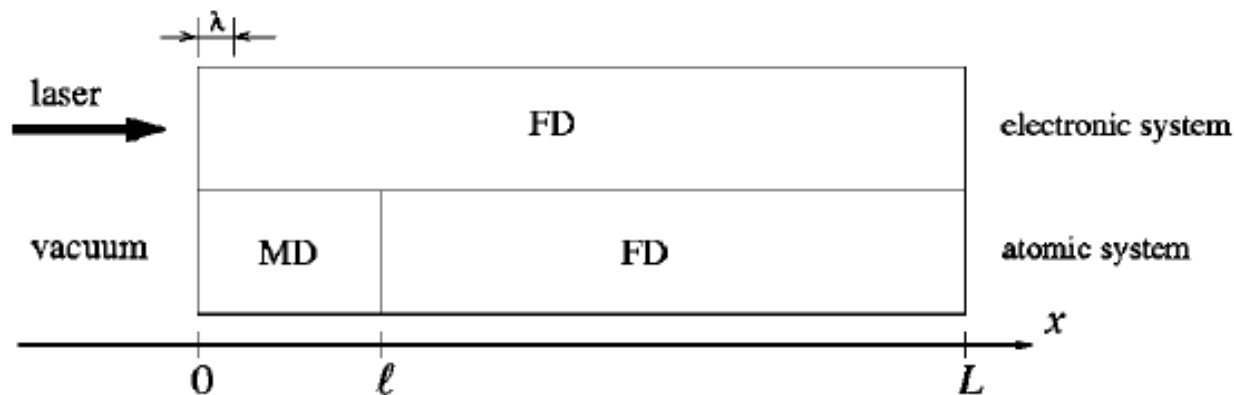
assumes electronic and atomic system to be internally thermalized with temperatures T_e , T_a

heat conduction equation for electrons
Newton's equations for atoms

electron-ion coupling terms

$$C_e(T_e) \frac{\partial T_e}{\partial t} = \frac{\partial}{\partial z} \left(\kappa \frac{\partial T_e}{\partial z} \right) - g \cdot (T_e - T_l) + S(z, t)$$

$$M \frac{d^2 \mathbf{r}_i}{dt^2} = -\nabla_{\mathbf{r}_i} V(\{\mathbf{r}_j\}) - \frac{g}{C_a} \frac{T_a - T_e}{T_a} M \frac{d\mathbf{r}_i}{dt},$$



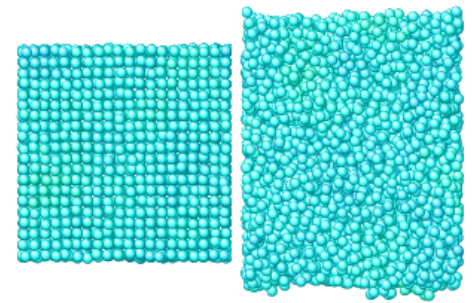
typical results for thin metal
films:

here: Al

with increasing energy input

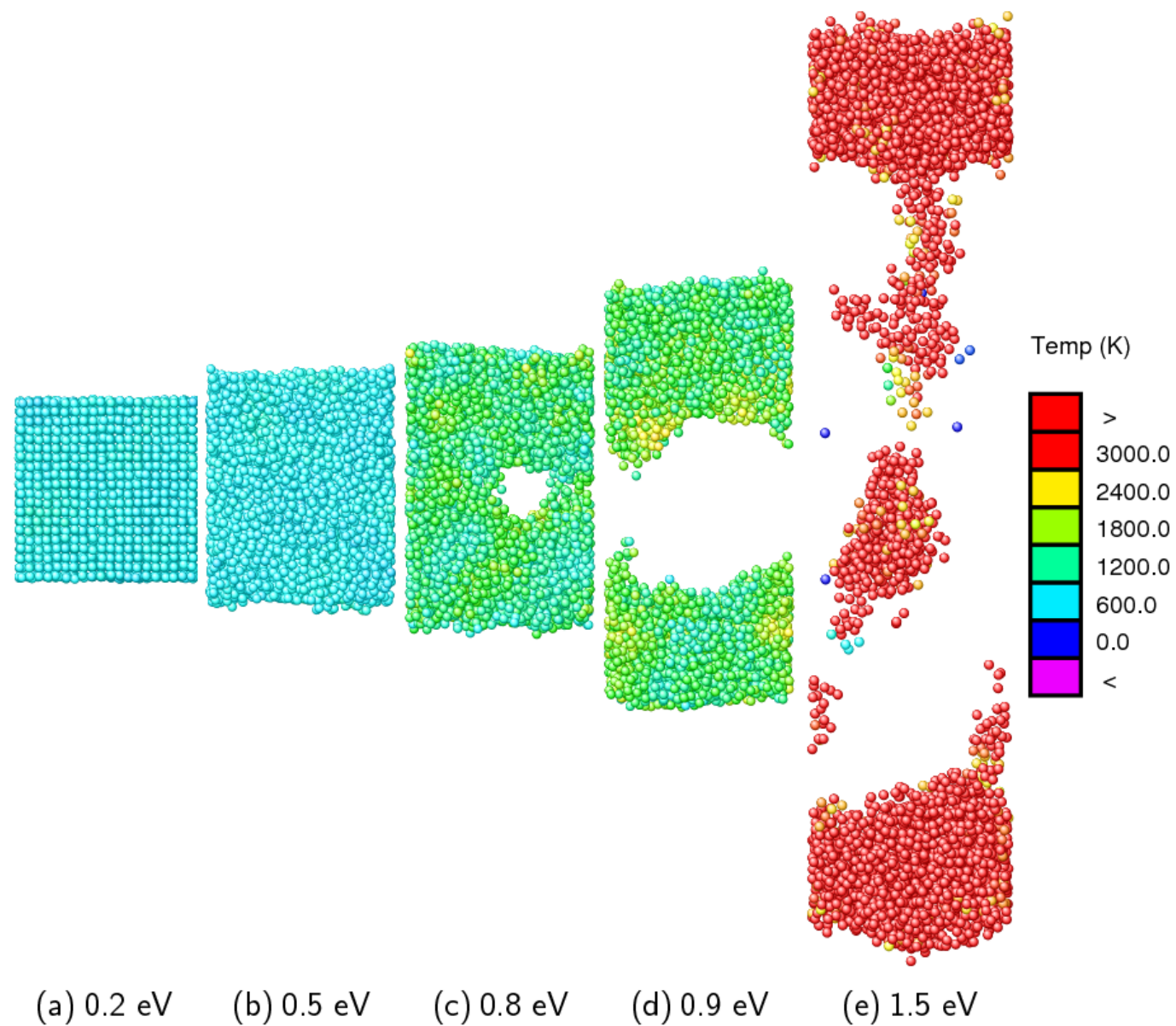
$E_0 = \text{absorbed energy} / \text{atom}$

melting ...



(a) 0.2 eV (b) 0.5 eV

metal
(Al)
target



melting

spallation

multi-fragmentation

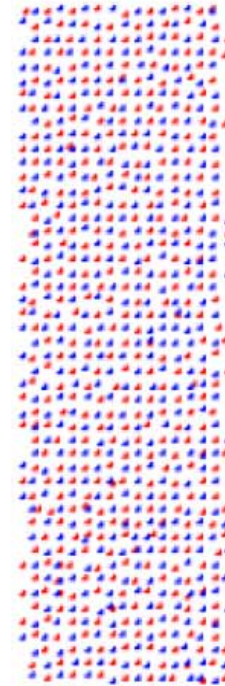
Outline

- Two-temperature model / MD for metals
- **Two-temperature model / MD for LiF**
- Melting and spallation of thin LiF films
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System:

LiF thin Film (10 nm)
(100) surface
lateral size

X-ray pulse: 7 ps
photon energy 90 eV



MD:

Buckingham potential + dispersion forces

$$V_{ij}(r) = \frac{q_i q_j}{4\pi\epsilon_0 r} + A_{ij} \exp(-r/\lambda_{ij}) - \frac{C_{ij}}{r^6}$$

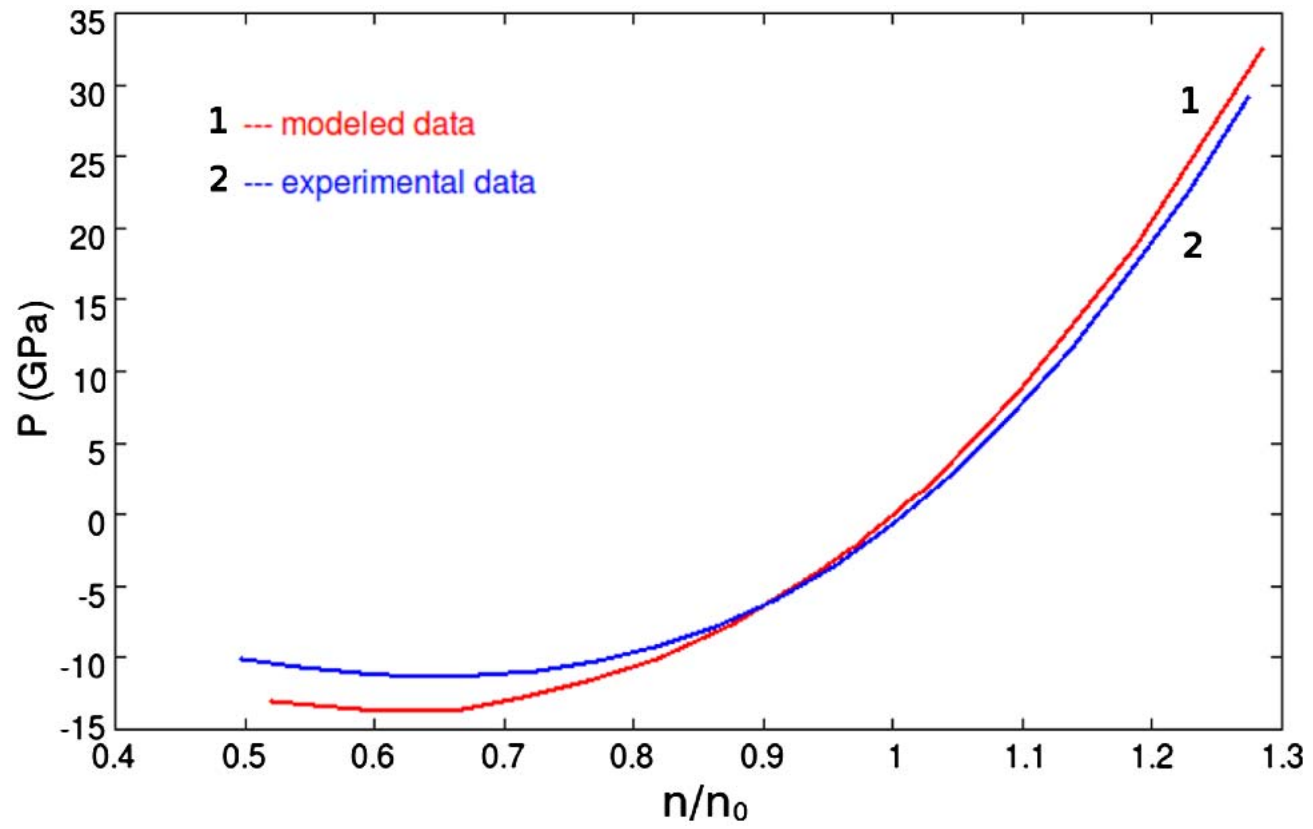
describes well:

elastic constants

yield strength

melting temperature ($T_m = 1118$ K)

...



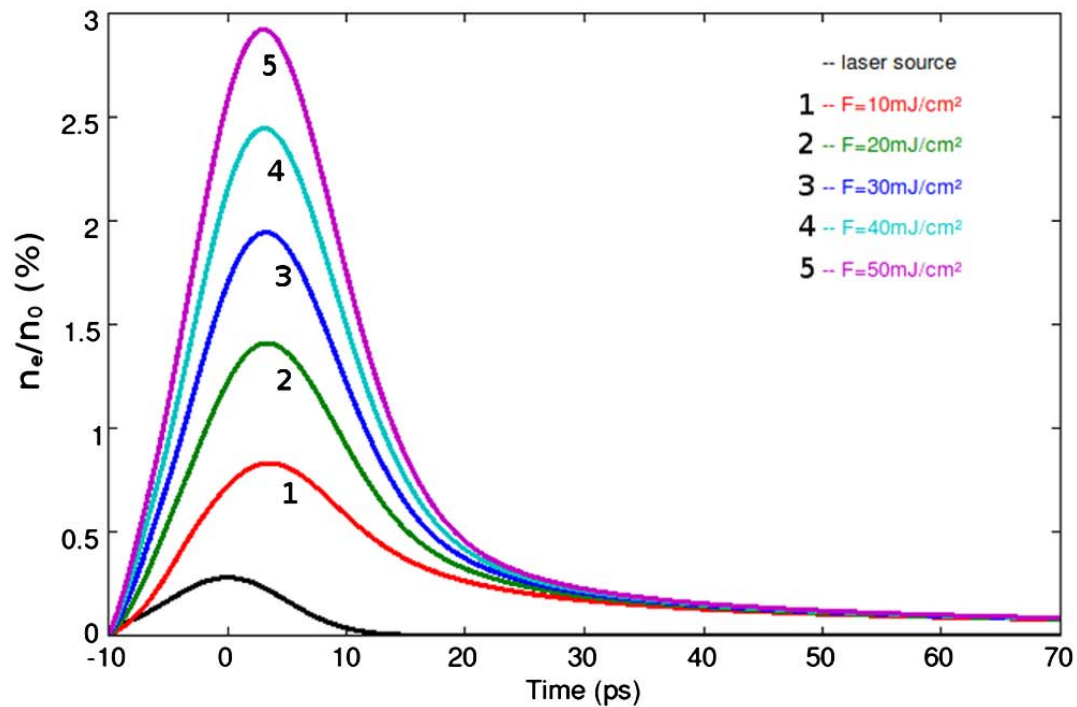
cold curve ($T=0$)

LiF
electron kinetics

n_e : electron density in conduction band

$$\frac{dn_e}{dt} = \frac{Q}{E_{\text{gap}}} + v_{\text{imp}} n_e - \kappa_{\text{rec}} n_e^3$$

$Q(t)$: laser source
 v_{imp} : impact ionization
 κ_{rec} : recombination
 E_{gap} : gap energy



Result:

Electron concentration
< 3 %

LiF electron kinetics

n_e : electron density in conduction band

$$\frac{dn_e}{dt} = \frac{Q}{E_{\text{gap}}} + v_{\text{imp}}n_e - \kappa_{\text{rec}}n_e^3$$

$Q(t)$: laser source
 v_{imp} : impact ionization
 κ_{rec} : recombination
 E_{gap} : gap energy

E_e : electron energy in conduction band

$$\frac{dE_e}{dt} = Q - \left(\frac{dE}{dt} \right)_{e \rightarrow a} = Q - AE_e$$

A : energy transfer in
electron-atom
collisions

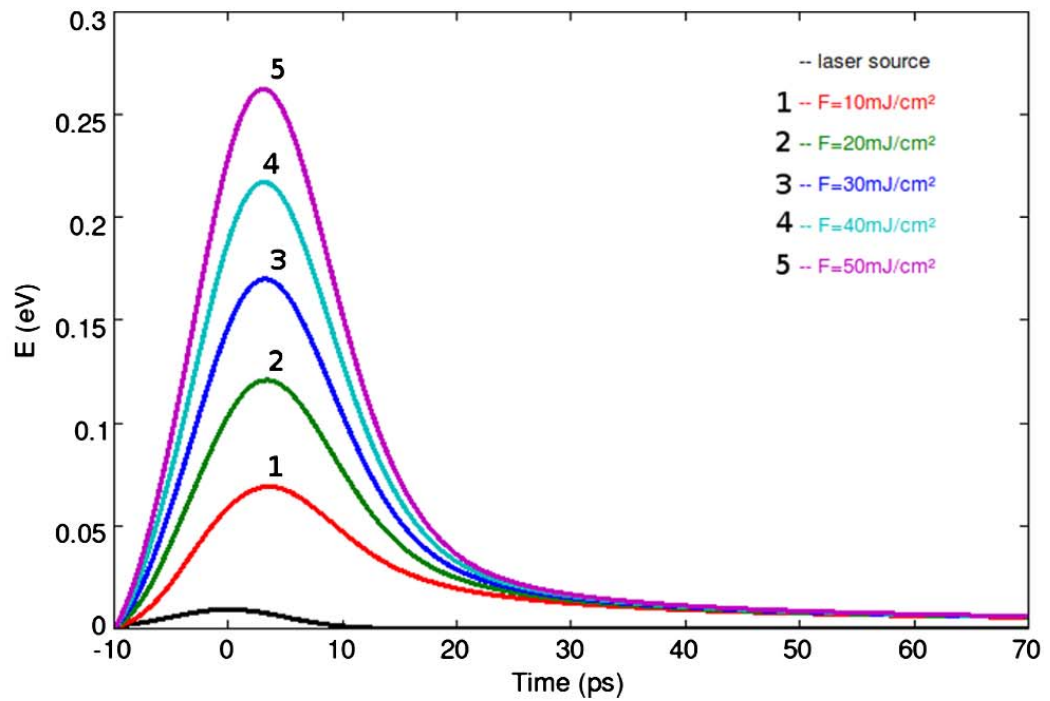
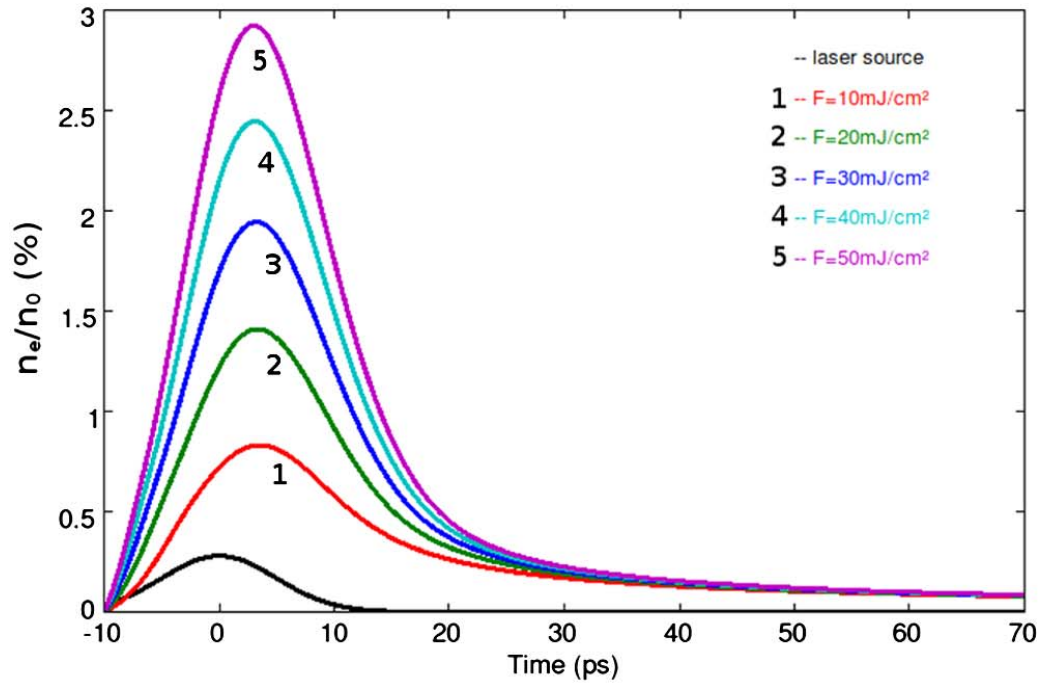
T_e : electron temperature

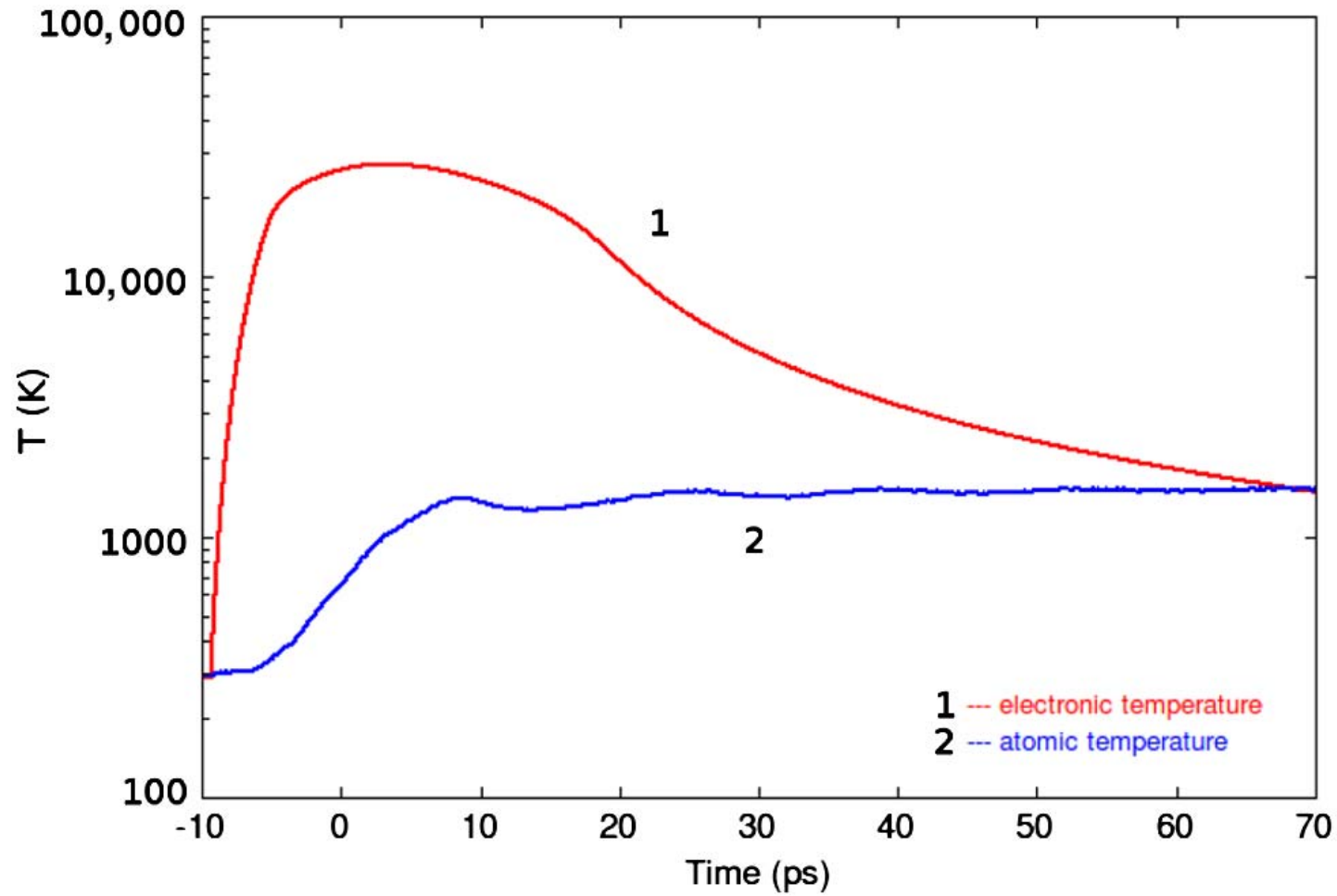
$A = 1 / \text{ps}$

$$E_e = n_e E_{\text{gap}} + E_{e,\text{kin}} = n_e E_{\text{gap}} + \frac{3}{2} n_e k T_e$$

Inogamov et al 2009

electron kinetics



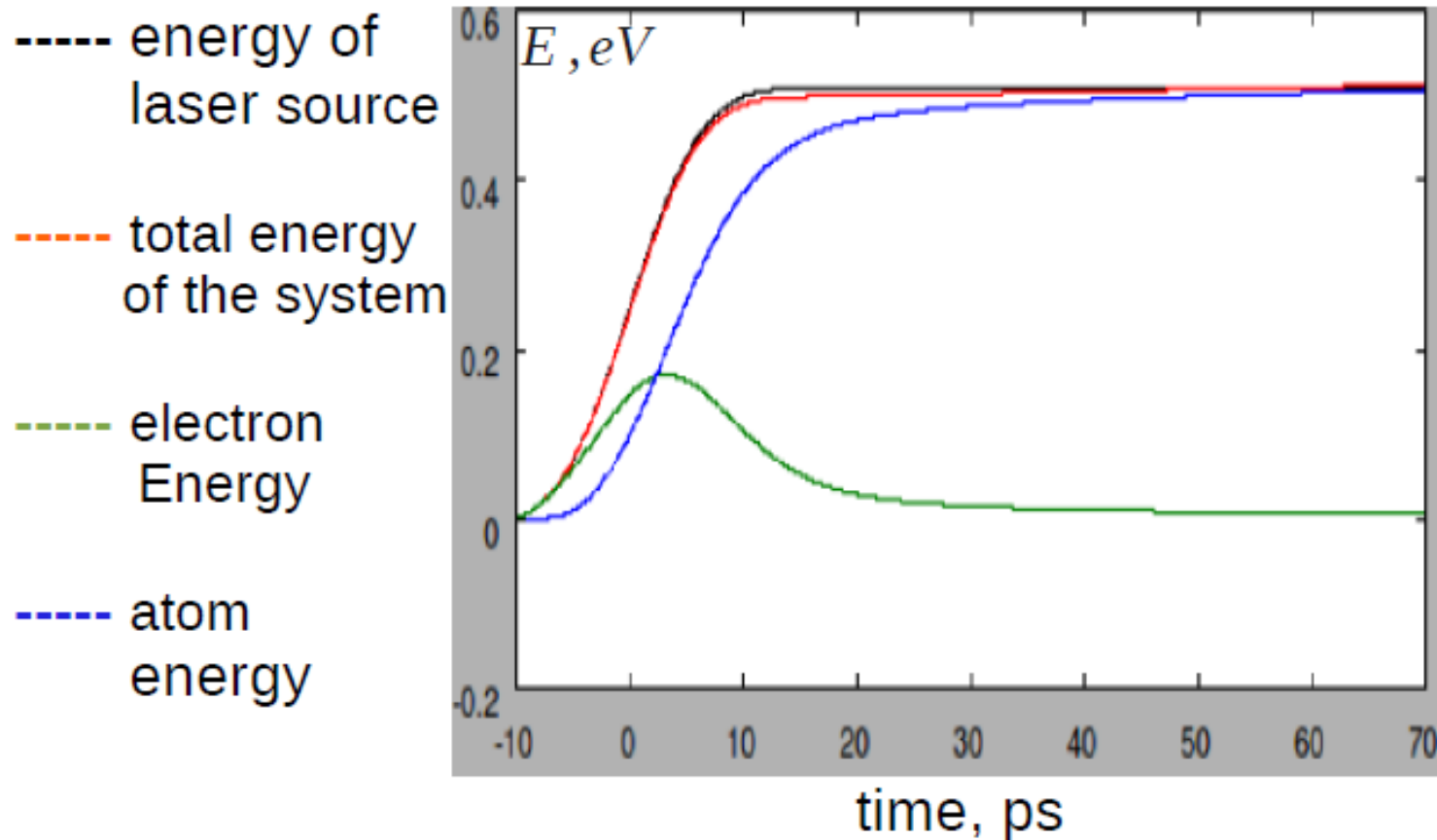


Electron and atom temperatures
after $F = 30 \text{ mJ/cm}^2$

LiF

coupling of electron kinetics and molecular dynamics

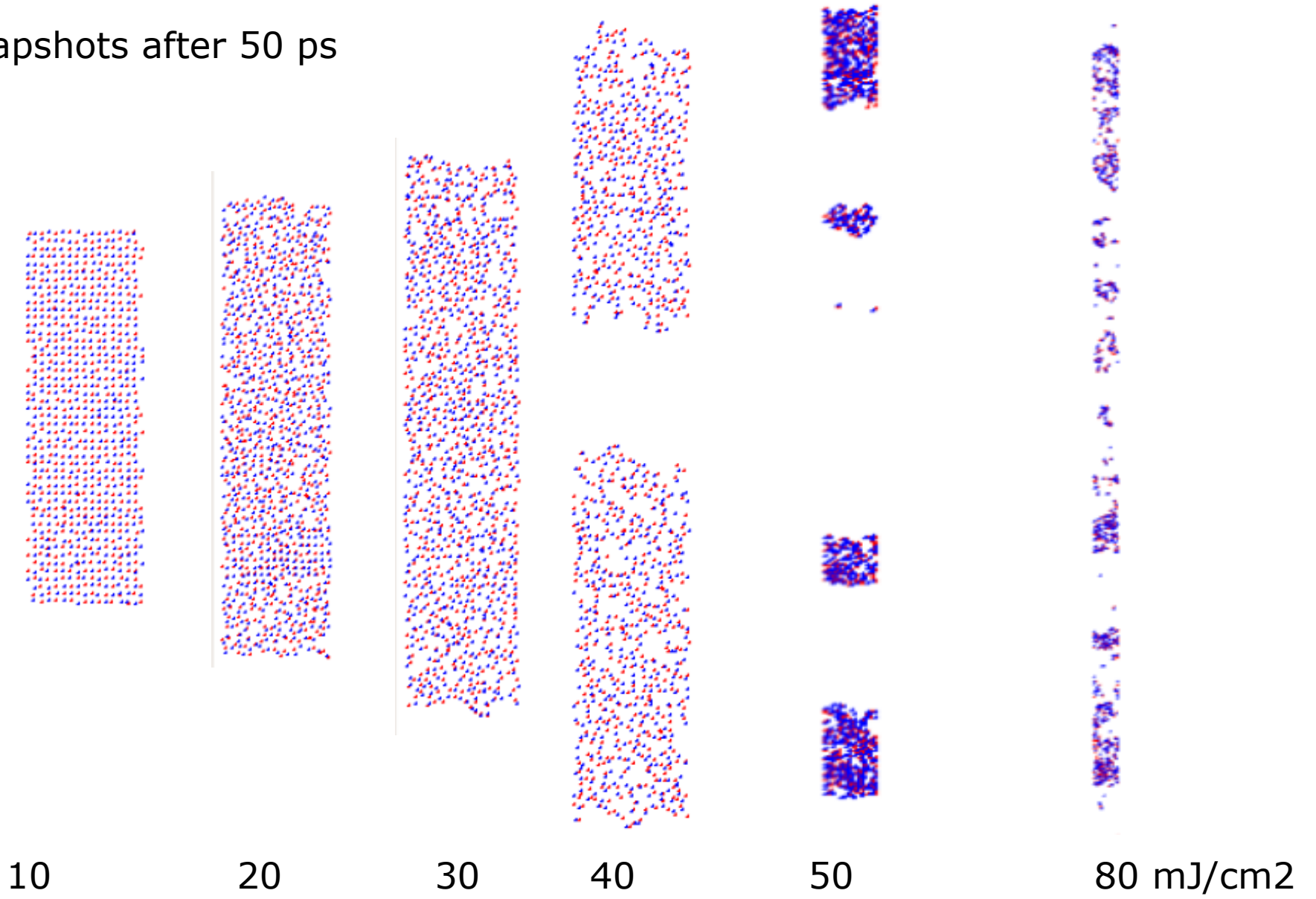
Energy conservation (per atom)

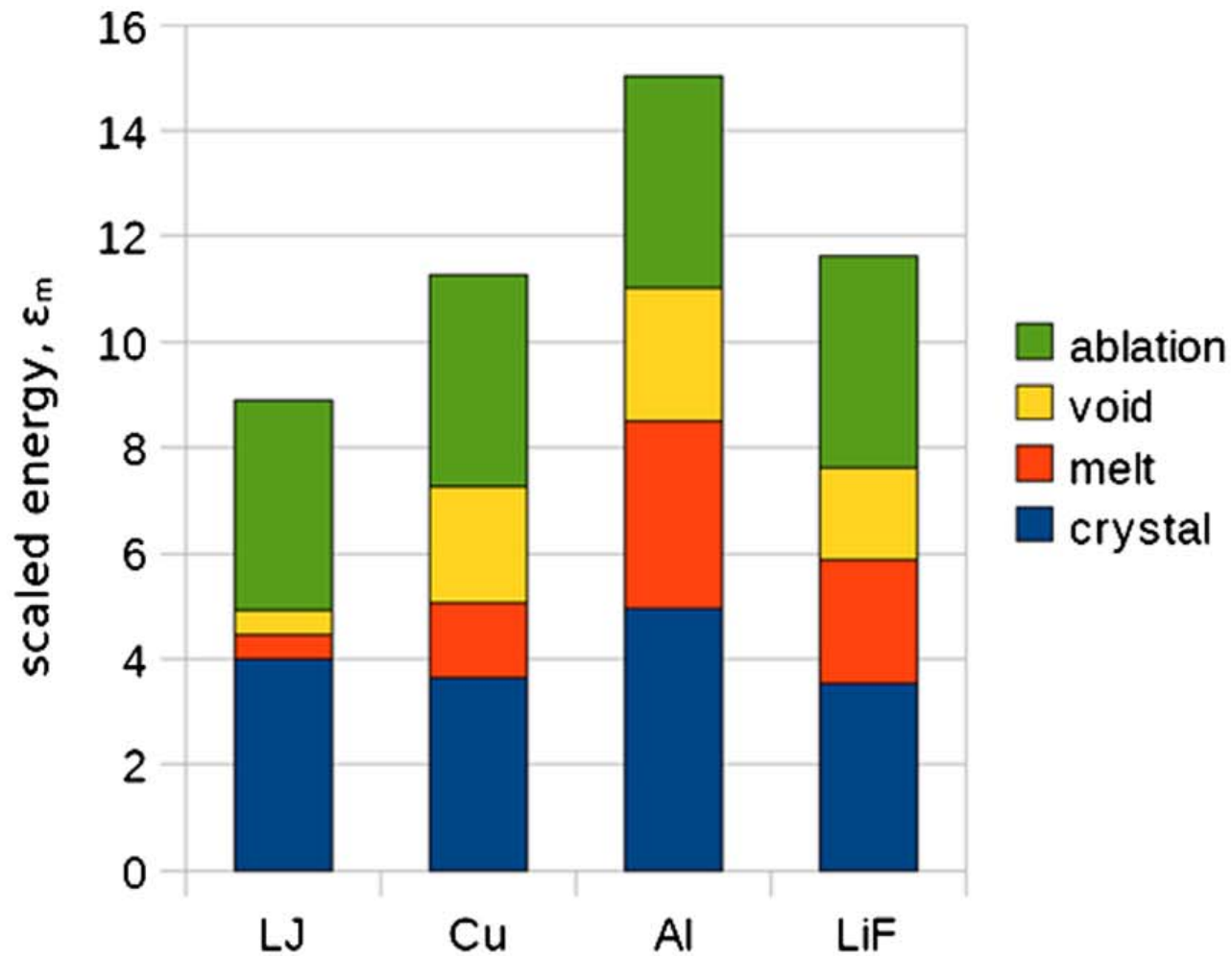


Outline

- Two-temperature model / MD for metals
- Two-temperature model / MD for LiF
- **Melting and spallation of thin LiF films**
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LiF
snapshots after 50 ps





Synopsis: threshold energies for various material classes

Absorbed energy / atom is scaled to melting temperature: E_0/kT_m

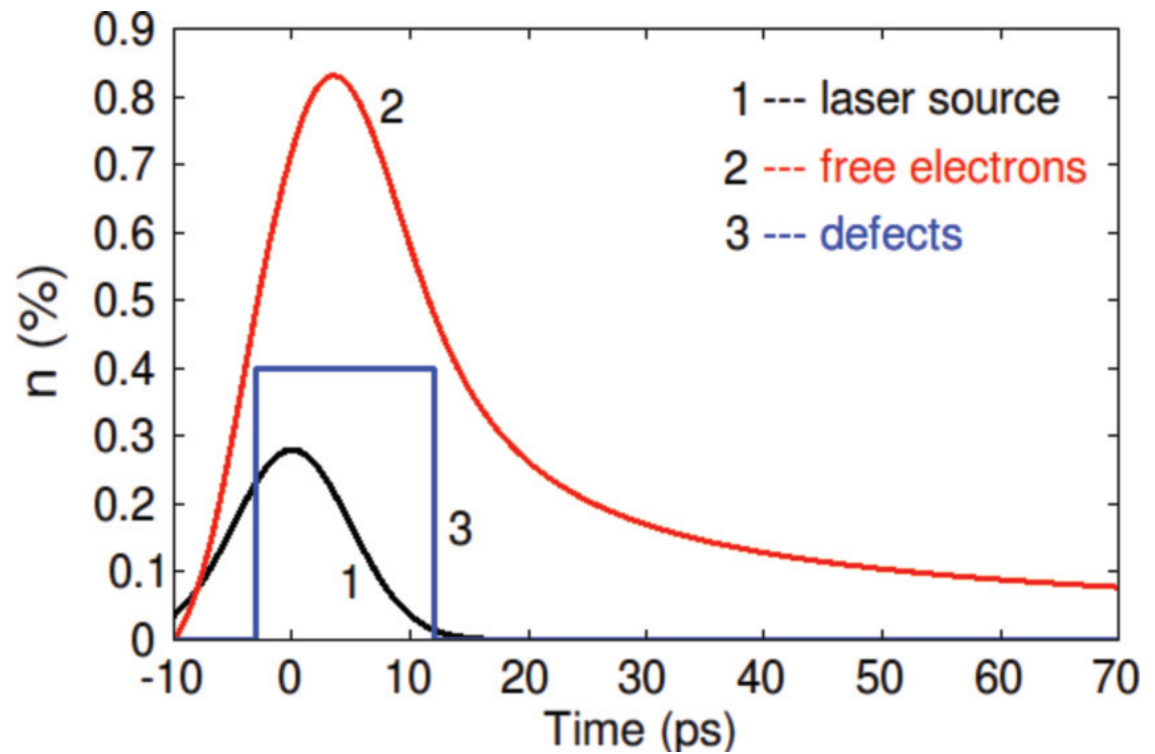
Outline

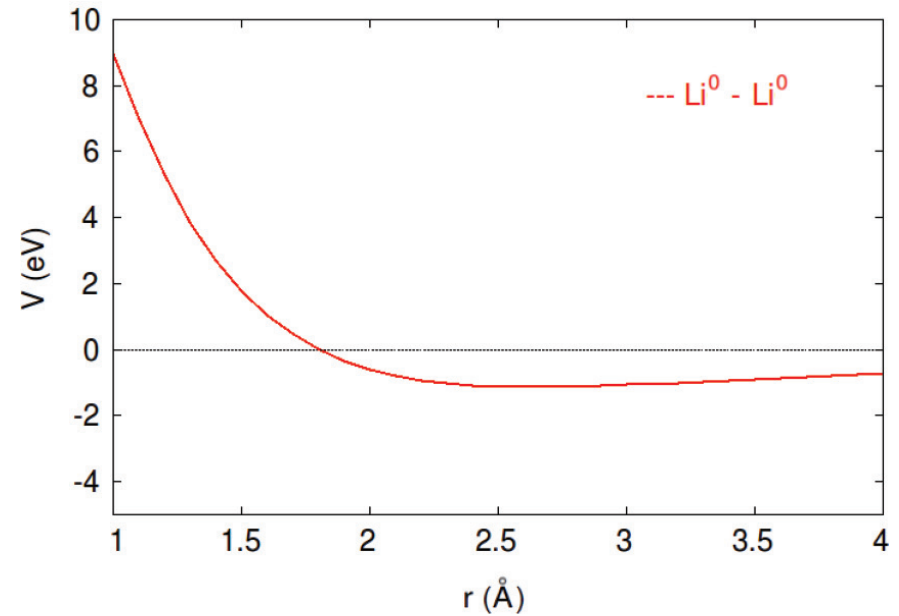
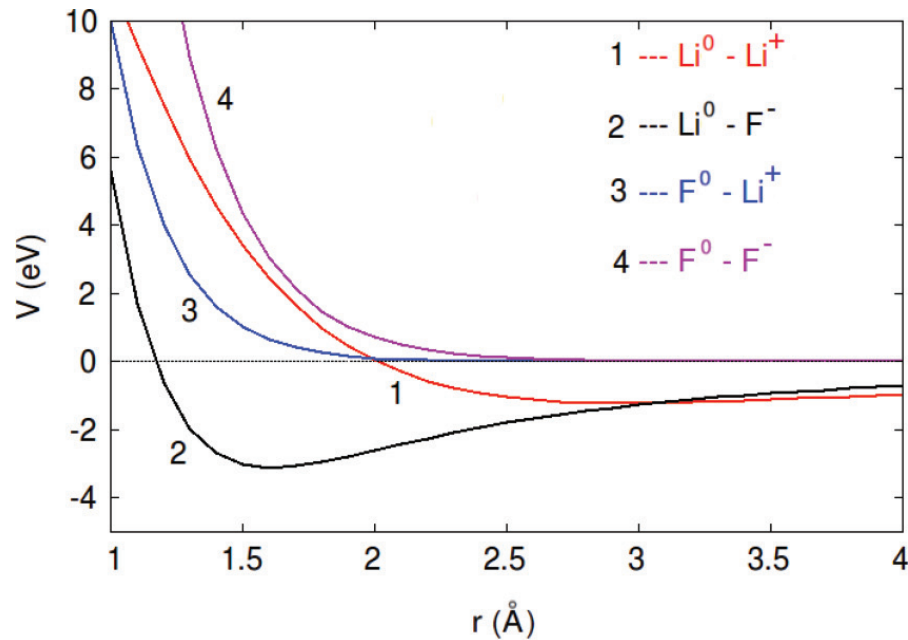
- Two-temperature model / MD for metals
- Two-temperature model / MD for LiF
- Melting and spallation of thin LiF films
- **defects**
- swift-ion tracks in LiF

Defects : neutral Li and F atoms



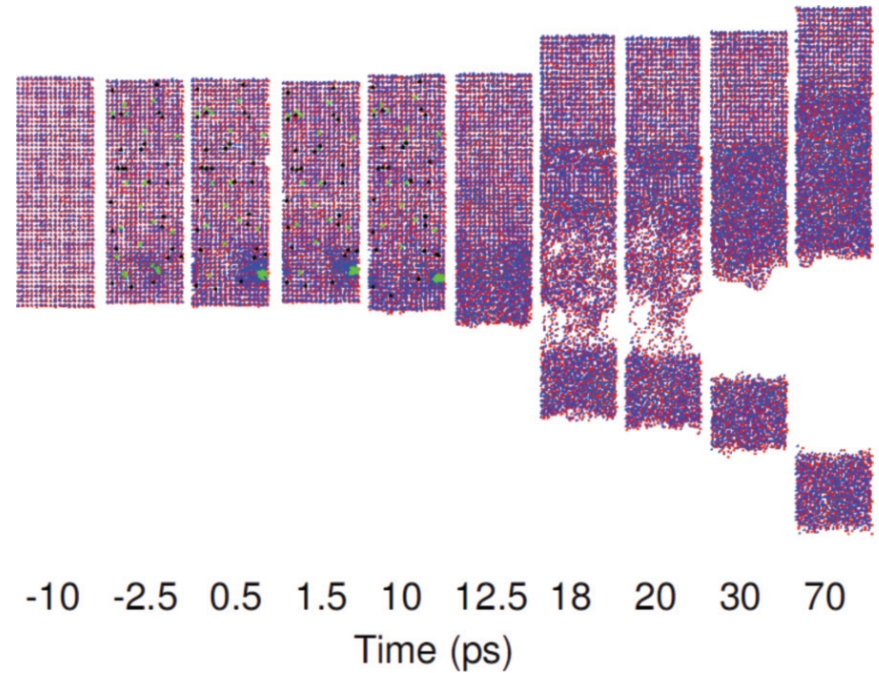
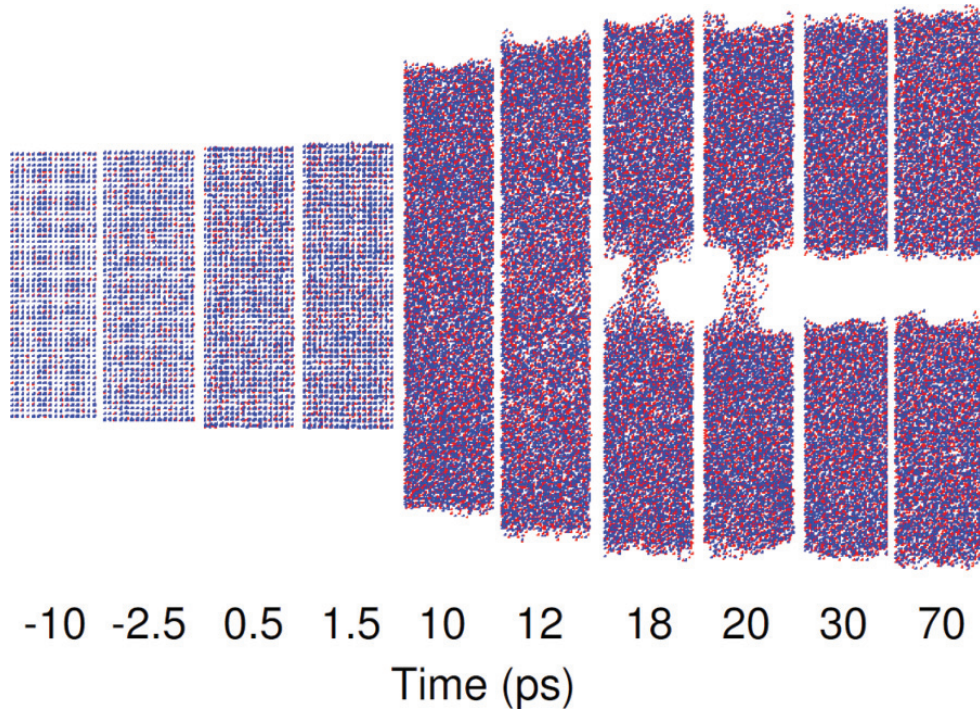
Defects are introduced *ad hoc*





Potentials taken from quantum chemical calculations:
[Wang et al, Phys Rev B 68 \(2003\) 115409](#)

Note: Li^0 is small \rightarrow may easily diffuse
 high polarizability \rightarrow attractive binding



Thermal ablation (no defects)

$$F = 38 \text{ mJ} / \text{cm}^2$$

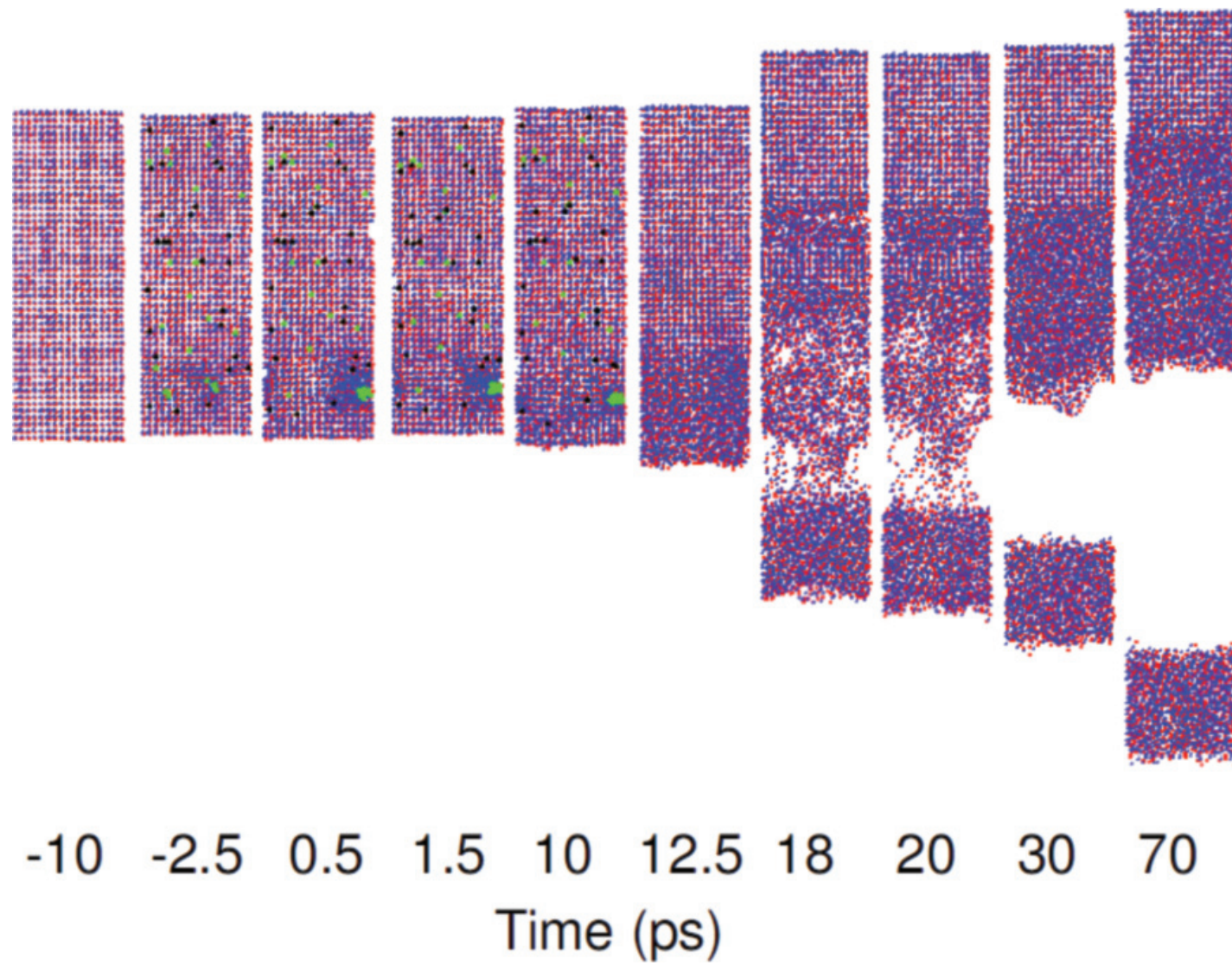
defect-supported ablation
(0.45% defects)

$$F = 10 \text{ mJ} / \text{cm}^2$$

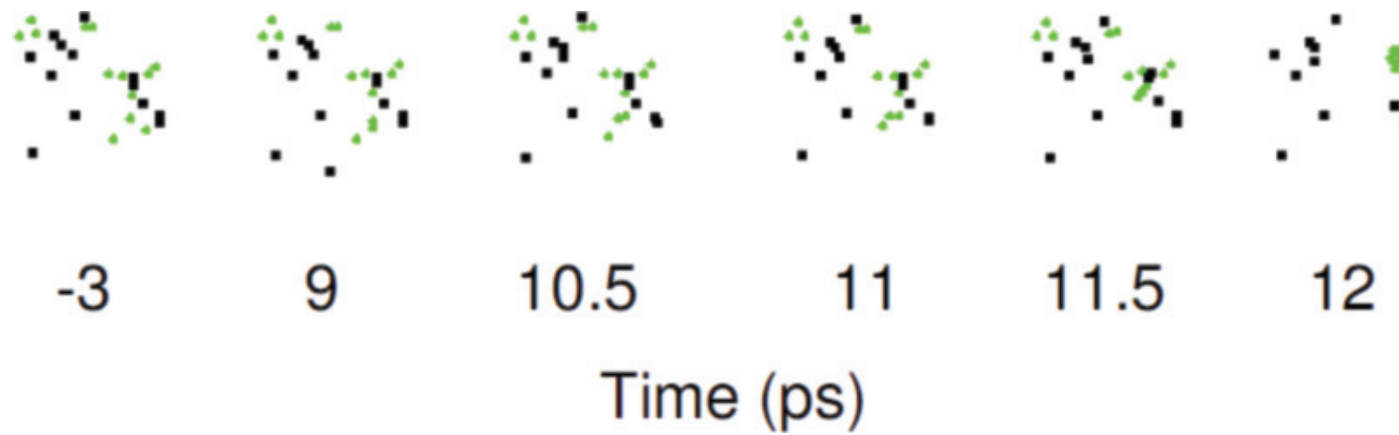
Result: defects lower ablation threshold

- bond weakening
- tensile pressure due to smaller atom radii

agrees with experiment



Green: Li cluster (metallic colloid)
destabilizes lattice



Top view of defects:
black: F^0 , green: Li^0

Formation of Li cluster by fast Li^0 diffusion
Condensation heat \rightarrow ablation

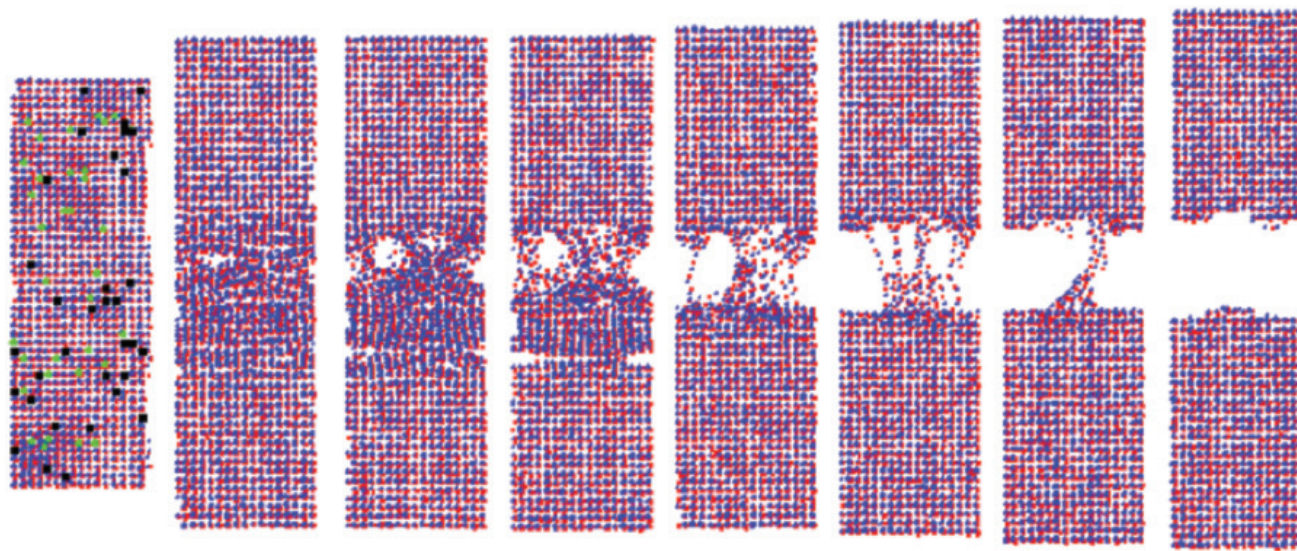
Experimentally observed under swift-particle irradiation of LiF

„Cold ablation“

Extreme case: assume that laser irradiation

- produces no free electrons (no target heating)
- only produces defects (potential energy)

Here defect concentration 0.57%



-2.5 14.25 14.75 15.25 16.25 18.5 21.5 41.5

Time (ps)

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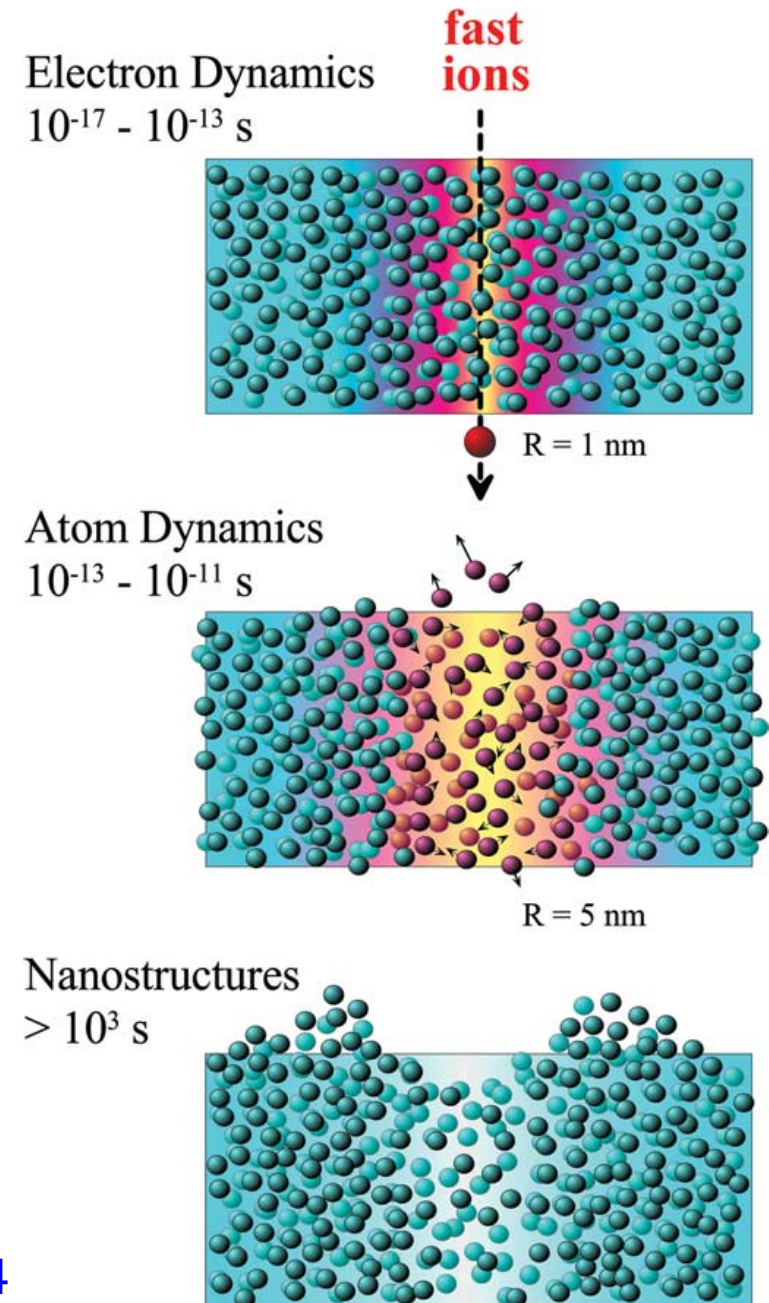
Swift ions deposit their energy
in the form of electronic excitation
in the target

„Similar physics“ as in
laser irradiation

How to treat system in MD:

After ion passage:

$F^- \rightarrow F^+ + 2e^-$ in track cylinder



Here:

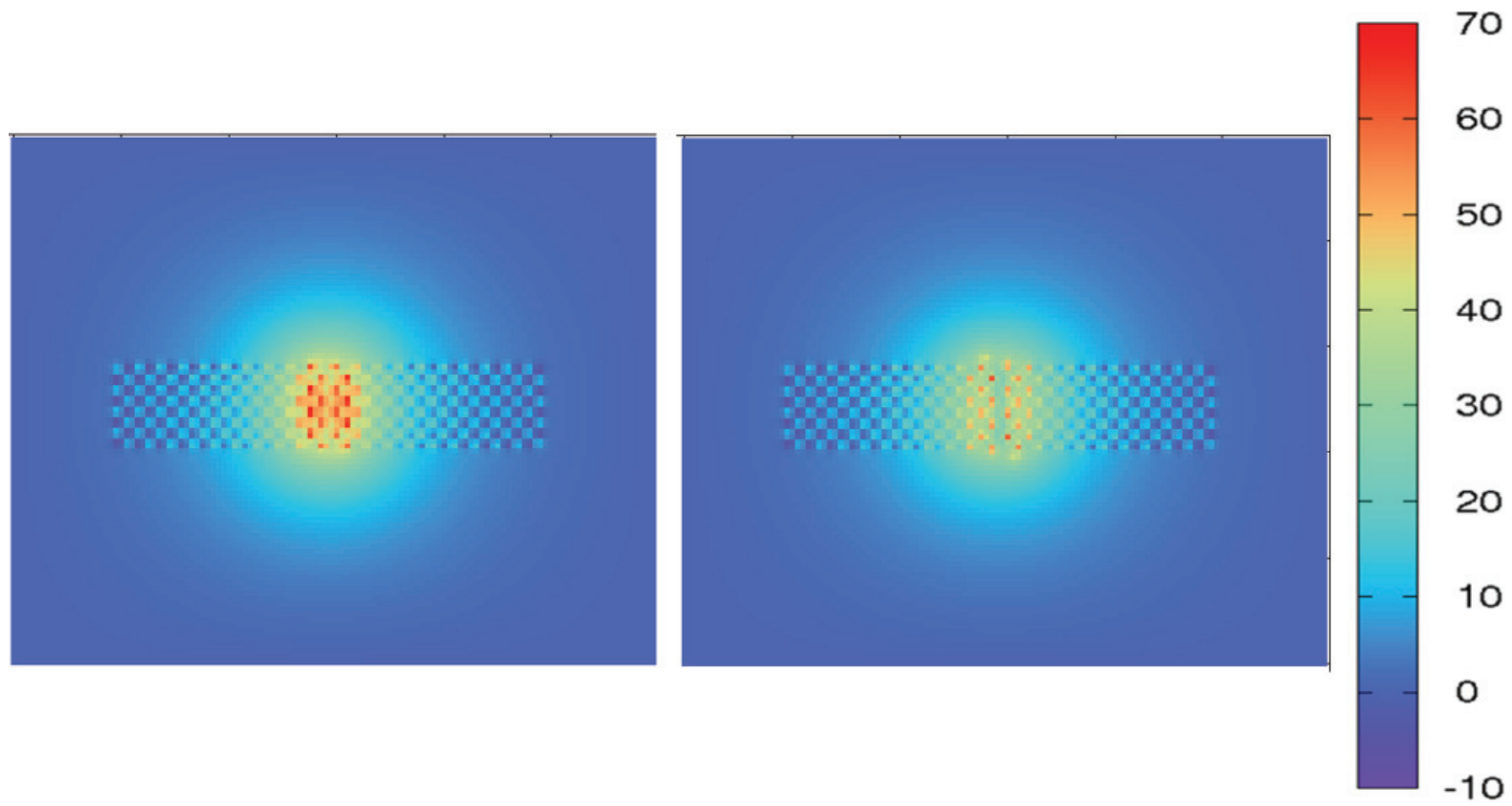
couple MD with a particle-in-cell (PIC) code for electron dynamics

Equations for electrons:

charge density: $\rho = e(Z_i n_i - n_e)$

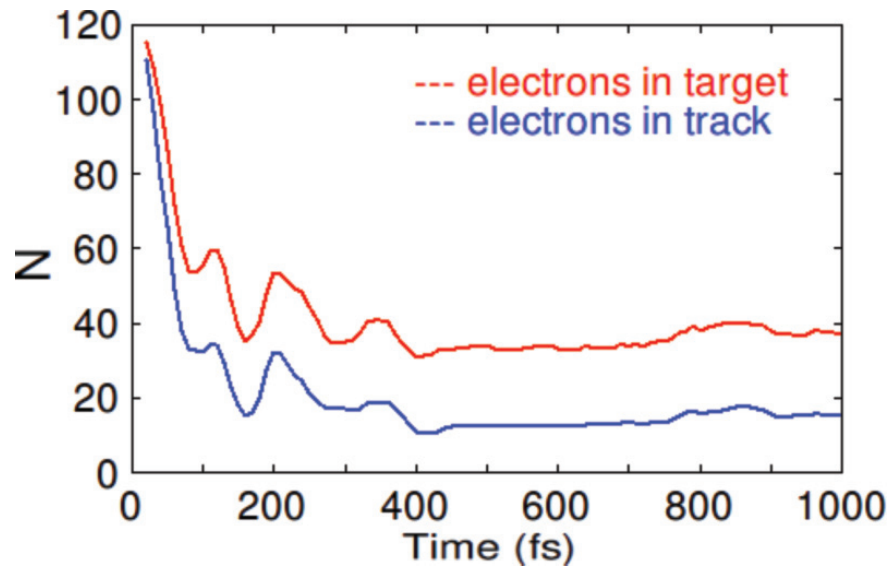
electric potential: $\nabla^2 \phi = -\frac{\rho}{\epsilon_0}$

number density: $n_e = n_0 \exp\left[\frac{e(\phi - \phi_0)}{k_B T_e}\right]$

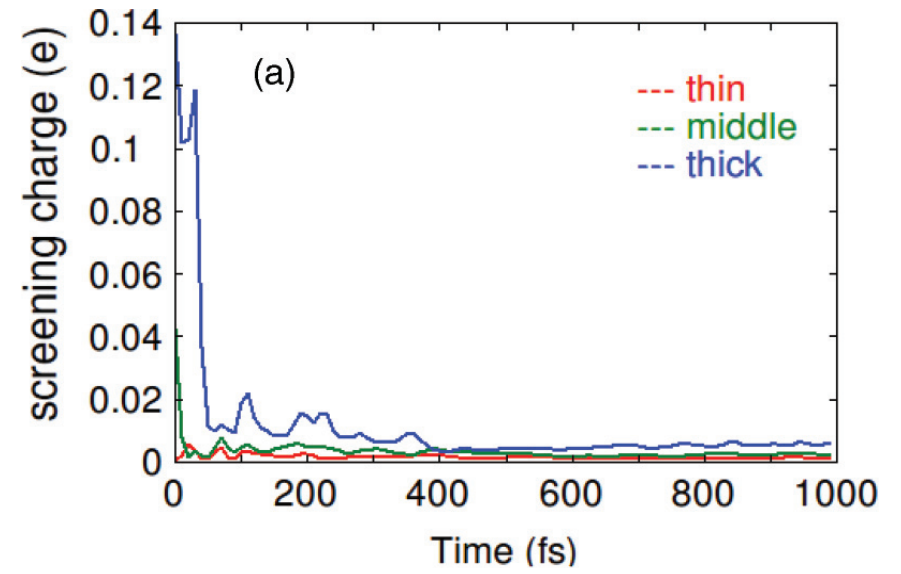


Electric potential:
at passage of ion

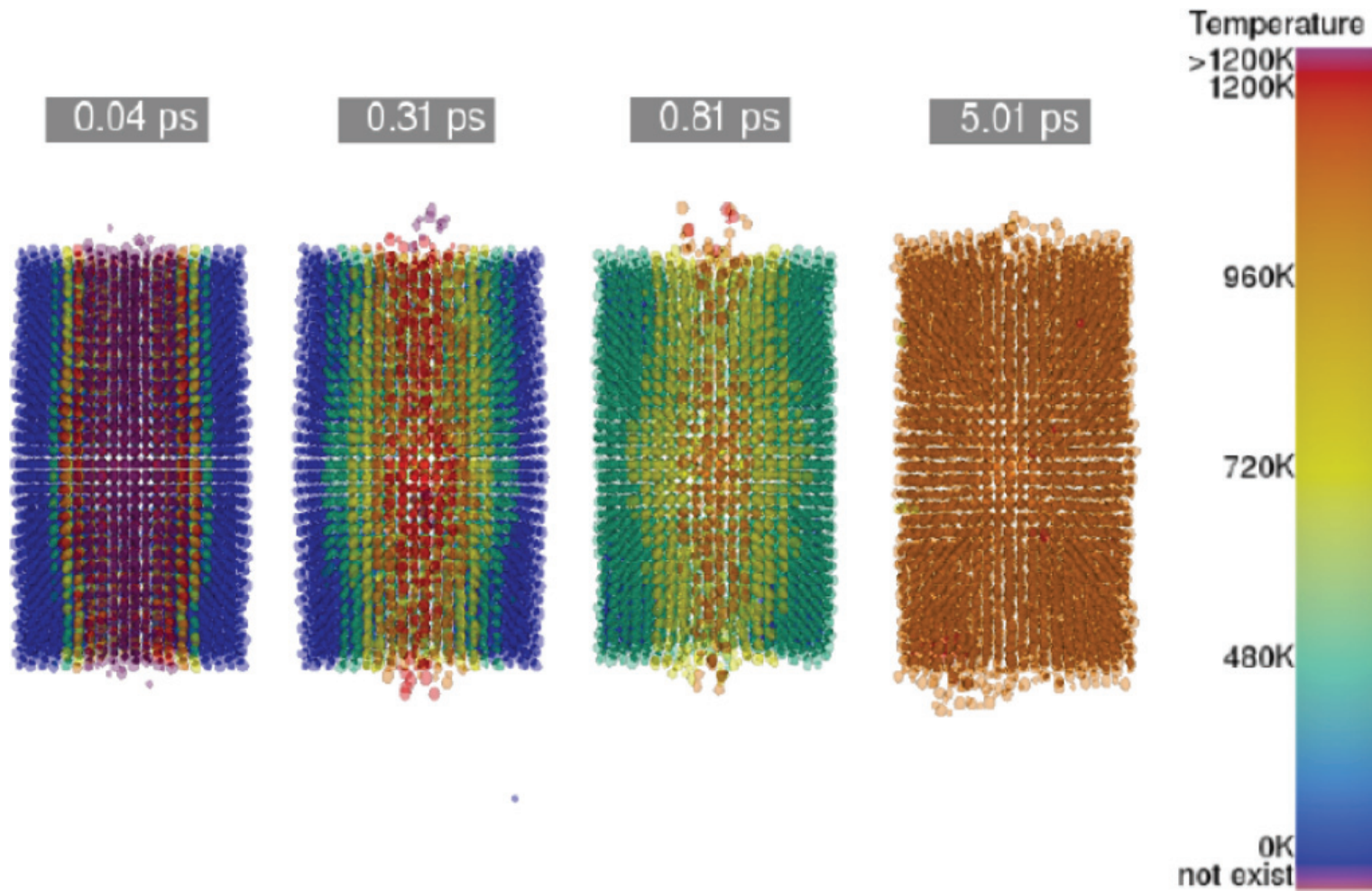
10 fs later



number of electrons remaining in track



shielding of F^+ ions



Evolution of ionization track: sputtering

Conclusions

Two-temperature model for LiF: need for plasma equations for electron density and energy

- Ablation mechanism similar as in metals: spallation in molten state
- Role of longlived defects
- Defects de-stabilize lattice -> lower ablation threshold
- even „cold“ ablation is possible
- role of metallic clusters:
 - form due to high Li^0 diffusion rate
 - destabilize lattice due to condensation heat
- efficient: potential energy introduced by defects small compared to laser energy (or thermal energy of electrons)

Cherednikov et al., J. Opt. Soc. Am. B 28, 1817 (2011)

Cherednikov et al., Phys Rev B 88, 134109 (2013)

Cherednikov et al., Phys Rev B 87, 245424 (2013)

